

Geologic and Environmental Characteristics of Rare Earth Element Deposit Types Found in the United States

Bob Seal rseal@usgs.gov

Phil Verplanck plv@usgs.gov

Brad Van Gosen bvangose@usgs.gov

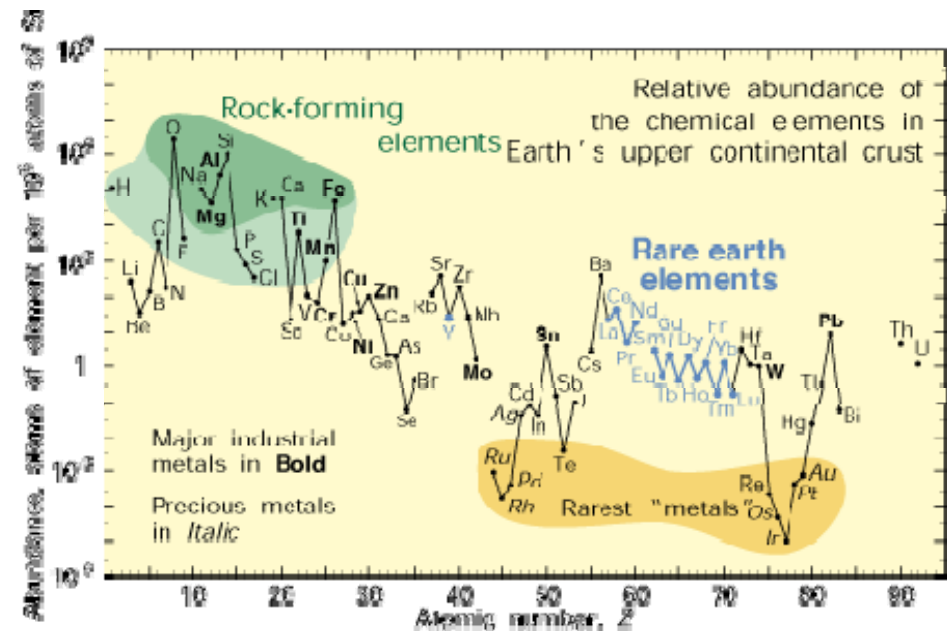
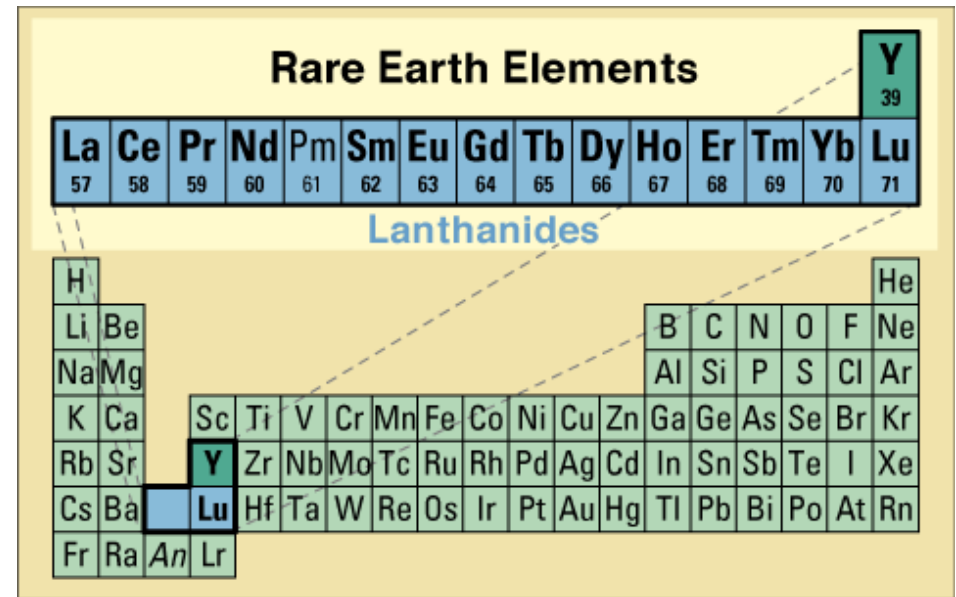
Andrew Grosz agrosz@usgs.gov

April 4, 2012

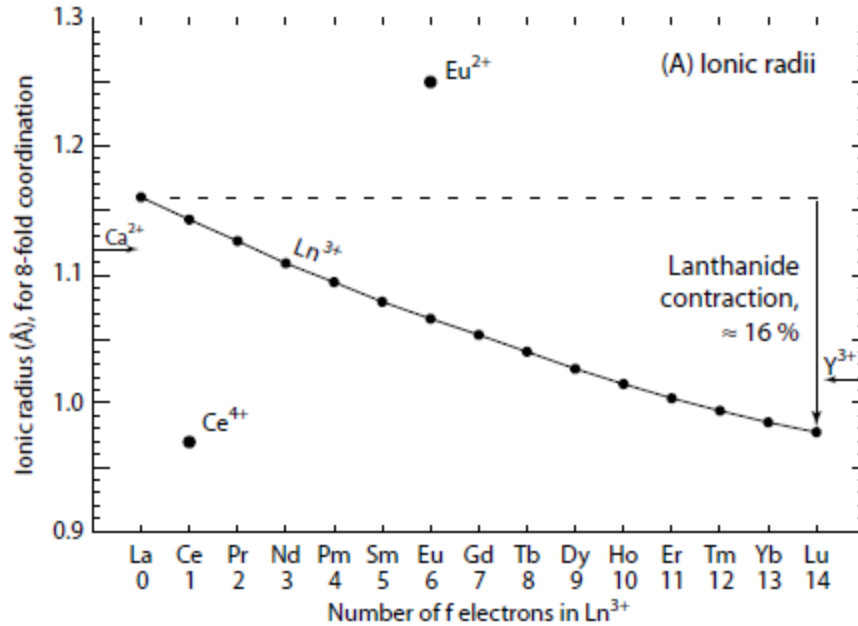


Rare Earth Elements

- Less abundant than common rock-forming elements
- More abundant than precious metals
- More or less abundant than base metals and metalloids.



Geochemical Controls in Nature

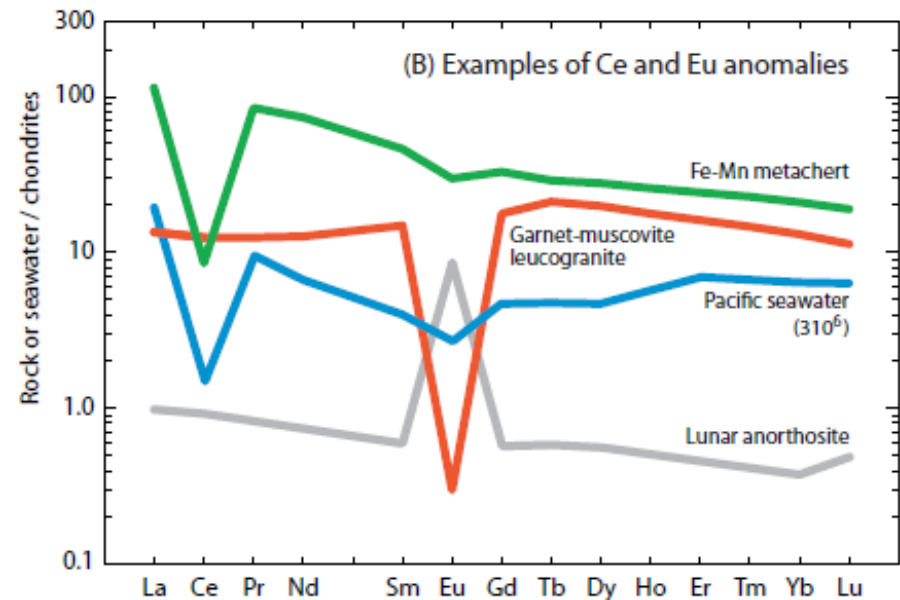


Lanthanide Contraction

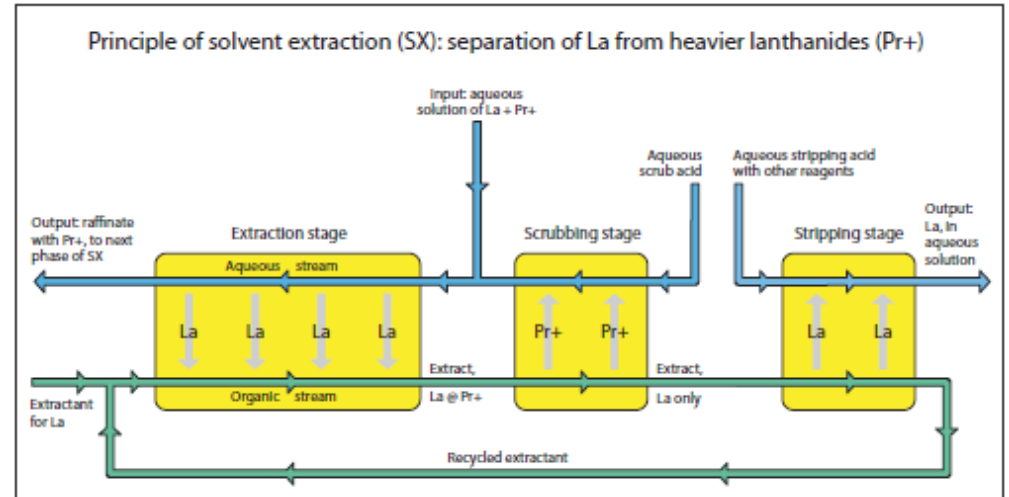
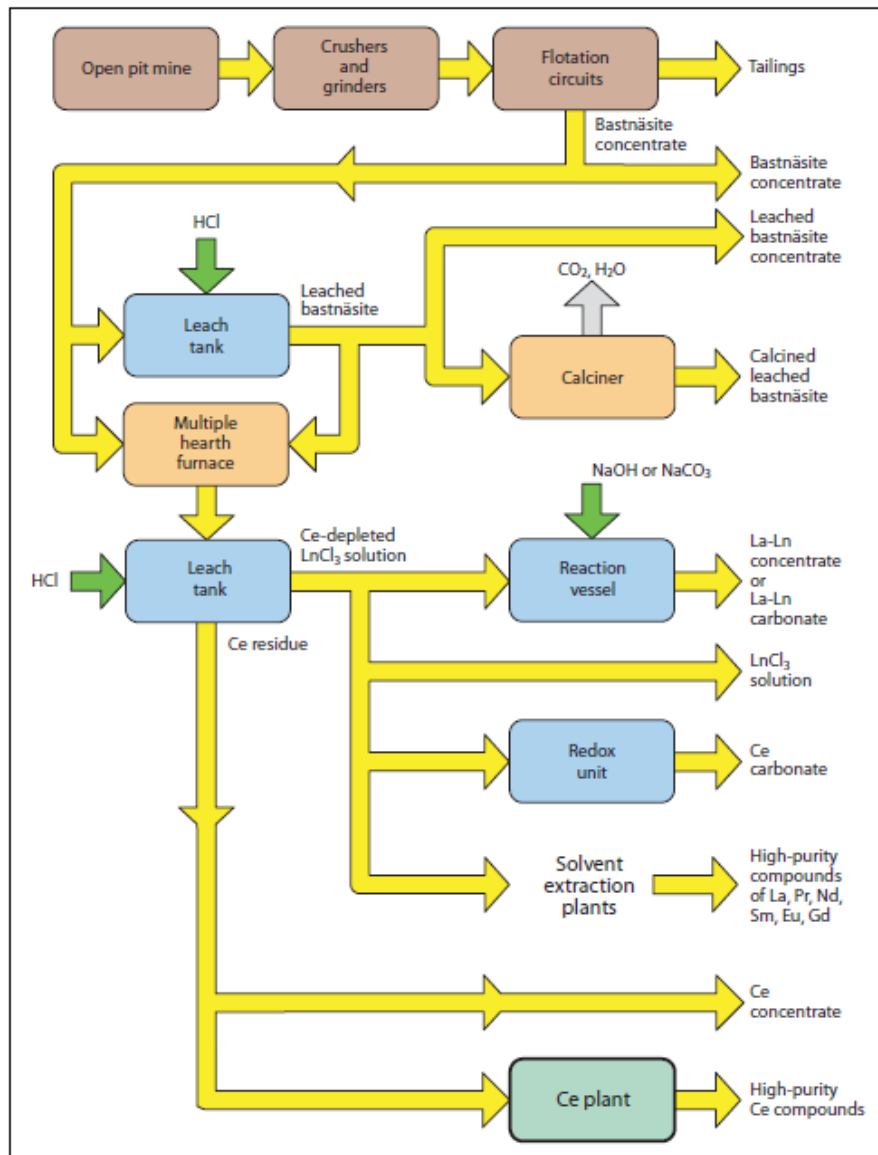
- Ionic radius decreases with increasing number

Oxidation State

- All REEs occur in 3+ state
- Eu also occurs in 2+ state
 - Substitutes for 2+ cations like Ca^{2+}
- Ce also occurs in 4+ state
 - Even more insoluble



Rare Earth Element Ore Processing



Recovery of individual REEs is complicated by:

- Similar geochemical behavior of REEs as a group
- Wide variety of REE host minerals

Primary and Byproduct REE Deposit Types

- Carbonatites
 - **Bayan Obo, China**
 - Mountain Pass, CA
- Alkaline Intrusion-Related Deposits
 - Bokan Mountain, AK
 - Thor Lake (Nechalacho), NWT, Canada
- Magmatic Magnetite-Hematite bodies
 - Pea Ridge Mine, Missouri
 - Olympic Dam, Australia?
 - Kiruna, Sweden?

Potential Byproduct REE Sources

- Monazite-bearing Placer Deposits
 - Central Idaho stream placers
 - North and South Carolina stream placers
 - Virginia, Georgia, and Florida beach deposits
- Phosphorite Deposits
 - Florida, North Carolina
 - Phosphoria Formation, ID, MT

Rare Earth Element Deposits of the United States



Rare Earth Element Minerals & Deposit Types

Group-Mineral	Formula	Carbonatite	Alkaline Intrusion-Related	Placer	Phosphorite
Oxides					
Aeschnite	$(\text{Ln,Ca,Fe})(\text{Ti,Nb})_2(\text{O,OH})_6$		X		
Euxenite	$(\text{Y,Ln,Ca})(\text{Nb,Ta,Ti})_2(\text{O,OH})_6$		X	X	
Fergusonite	YNbO_4		X		
Carbonates					
Bastnäsite	$(\text{Ln,Y})\text{CO}_3\text{F}$	X	X		
Parisite	$\text{Ca}(\text{Ln})_2(\text{CO}_3)_3\text{F}_2$	X	X		
Synchisite	$\text{Ca}(\text{Ln,Y})(\text{CO}_3)_2\text{F}$	X	X		
Tengerite	$\text{Y}_2(\text{CO}_3)_3 \cdot n(\text{H}_2\text{O})$		X		
Phosphates					
Apatite	$(\text{Ca,Ln})_5(\text{PO}_4)_3(\text{OH,F,Cl})$	X	X		X
Monazite	$(\text{Ln,Th})\text{PO}_4$	X	X	X	
Xenotime	YPO_4		X	X	
Silicates					
Allanite	$(\text{Ln,Y,Ca})_2(\text{Al,Fe}^{3+})_2(\text{SiO}_4)_3(\text{OH})$		X		
Eudialyte	$\text{Na}_4(\text{Ca,Ce})_2(\text{Fe}^{2+},\text{Mn}^{2+},\text{Y})\text{ZrSi}_8\text{O}_{22}(\text{OH,Cl})_2$		X		
Thalenite	$\text{Y}_2\text{Si}_2\text{O}_7$		X		
Zircon	$(\text{Zr,Ln})\text{SiO}_4$		X	X	

Carbonatite Deposits



Carbonatite Deposits

- Carbonatite – an igneous rock with greater than 50 % carbonate minerals (calcite, dolomite, ankerite)
- Less than 550 carbonatites known in world.
- Commonly associated with alkaline or peralkaline igneous rocks.
- The REE are “incompatible elements”, such as Th, Nb, and Zr, that do not tend to participate in the earlier mineral-forming events.
- Important sources of Nb, Ta, and REEs.

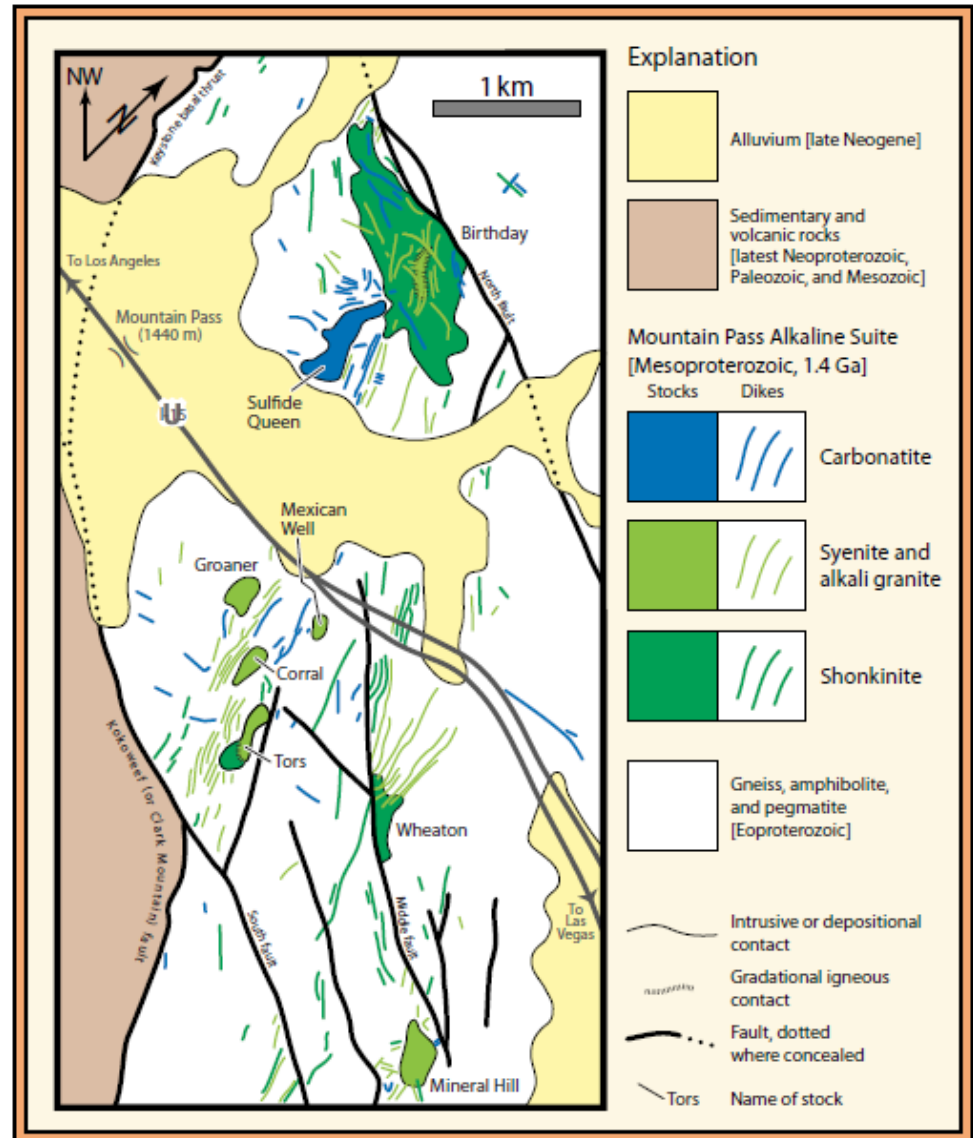
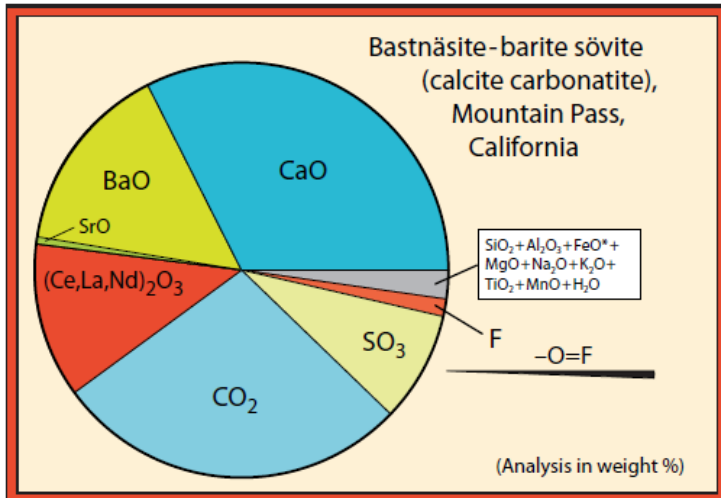
Bastnäsite
(REE)CO₃F



Mountain Pass, California

Mountain Pass, California

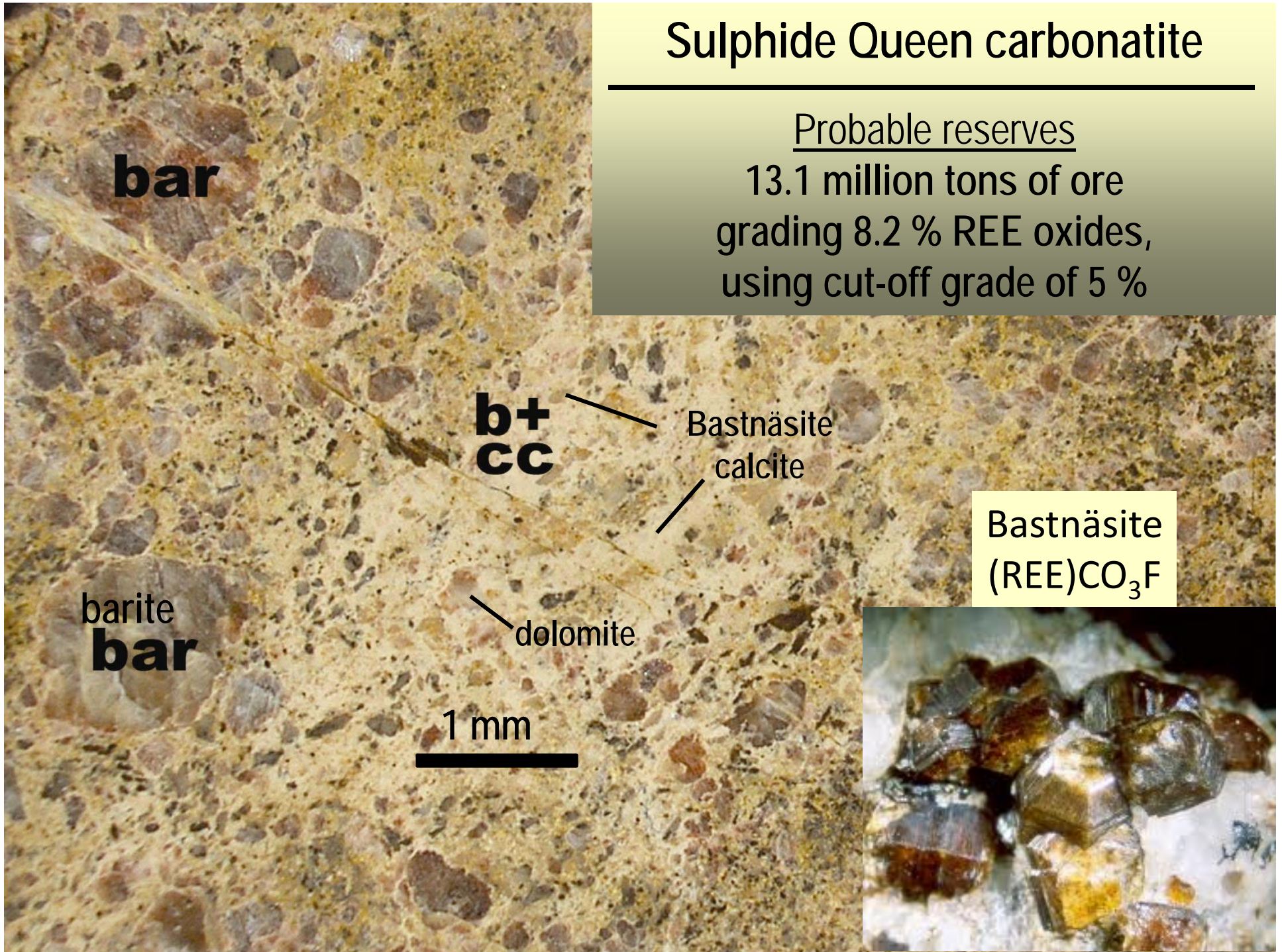
- Located in Mojave Desert
- Discovered in 1949 by uranium prospector.
- 1.4 Ga carbonatite complex
- Stopped mining in 2002
- Reopened 2010



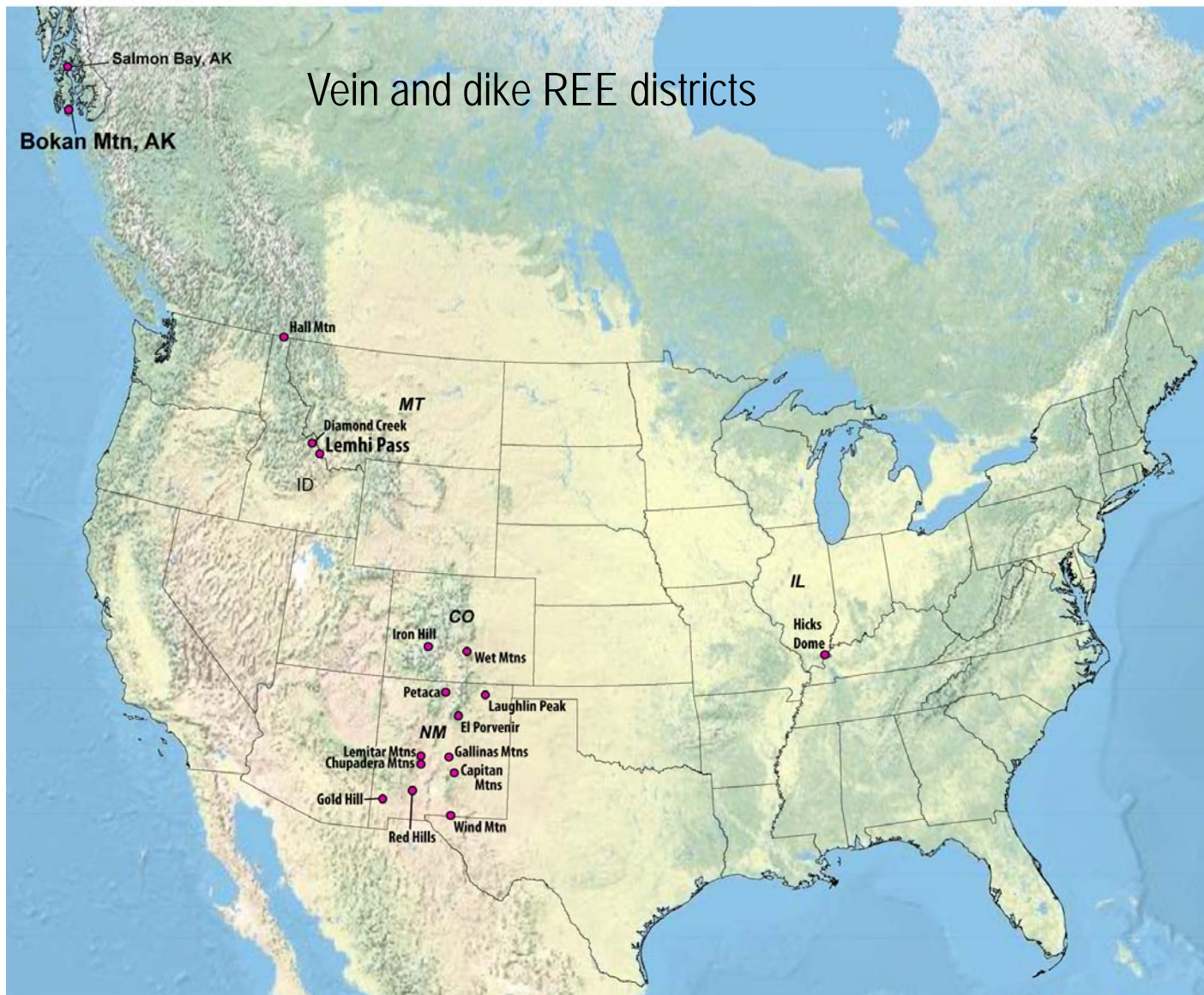
Sulphide Queen carbonatite

Probable reserves

13.1 million tons of ore
grading 8.2 % REE oxides,
using cut-off grade of 5 %



Alkaline Intrusion-Related Deposits

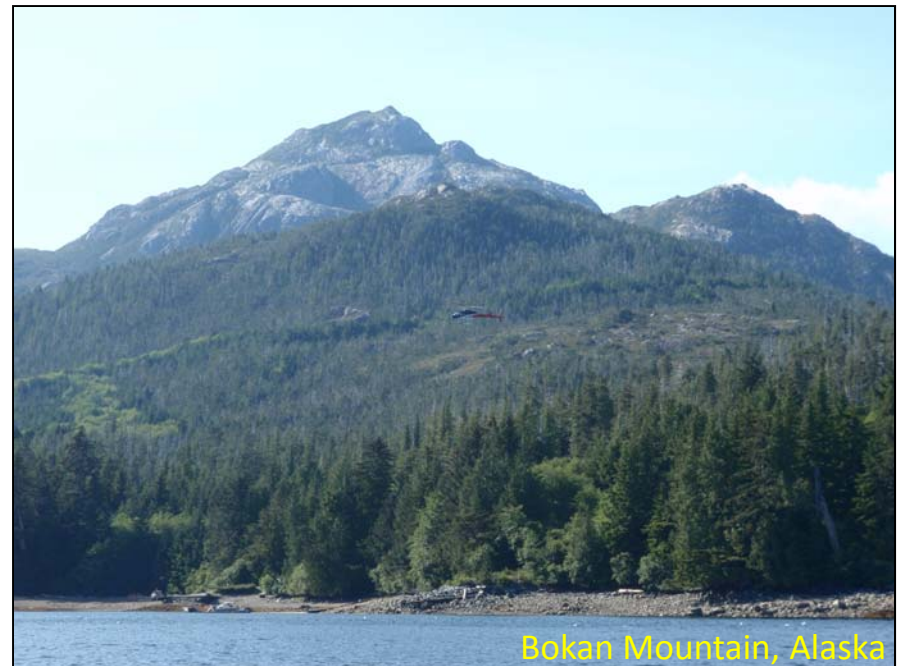


Alkaline Intrusion-Related Deposits

- Associated with alkaline or peralkaline igneous complexes

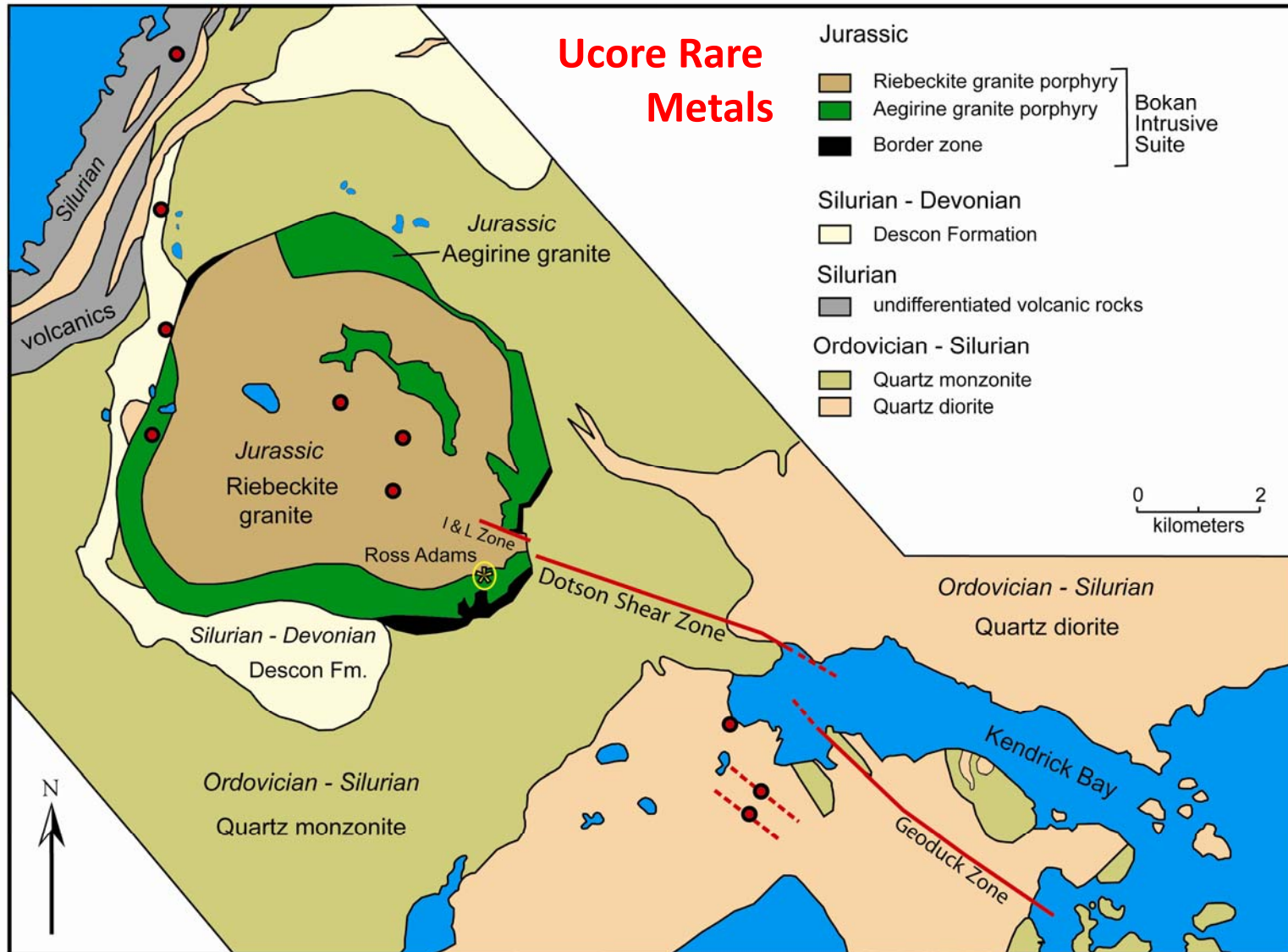
Peralkaline: $\text{Na}_2\text{O} + \text{K}_2\text{O} > \text{Al}_2\text{O}_3$

- Structural control on the veins, dikes & breccias
- In zoned alkaline intrusive complexes, REE veins, dikes, and breccia bodies are late phases.
- The REE are “incompatible elements”, such as Th, Nb, and Zr, that do not tend to participate in the earlier mineral-forming events.

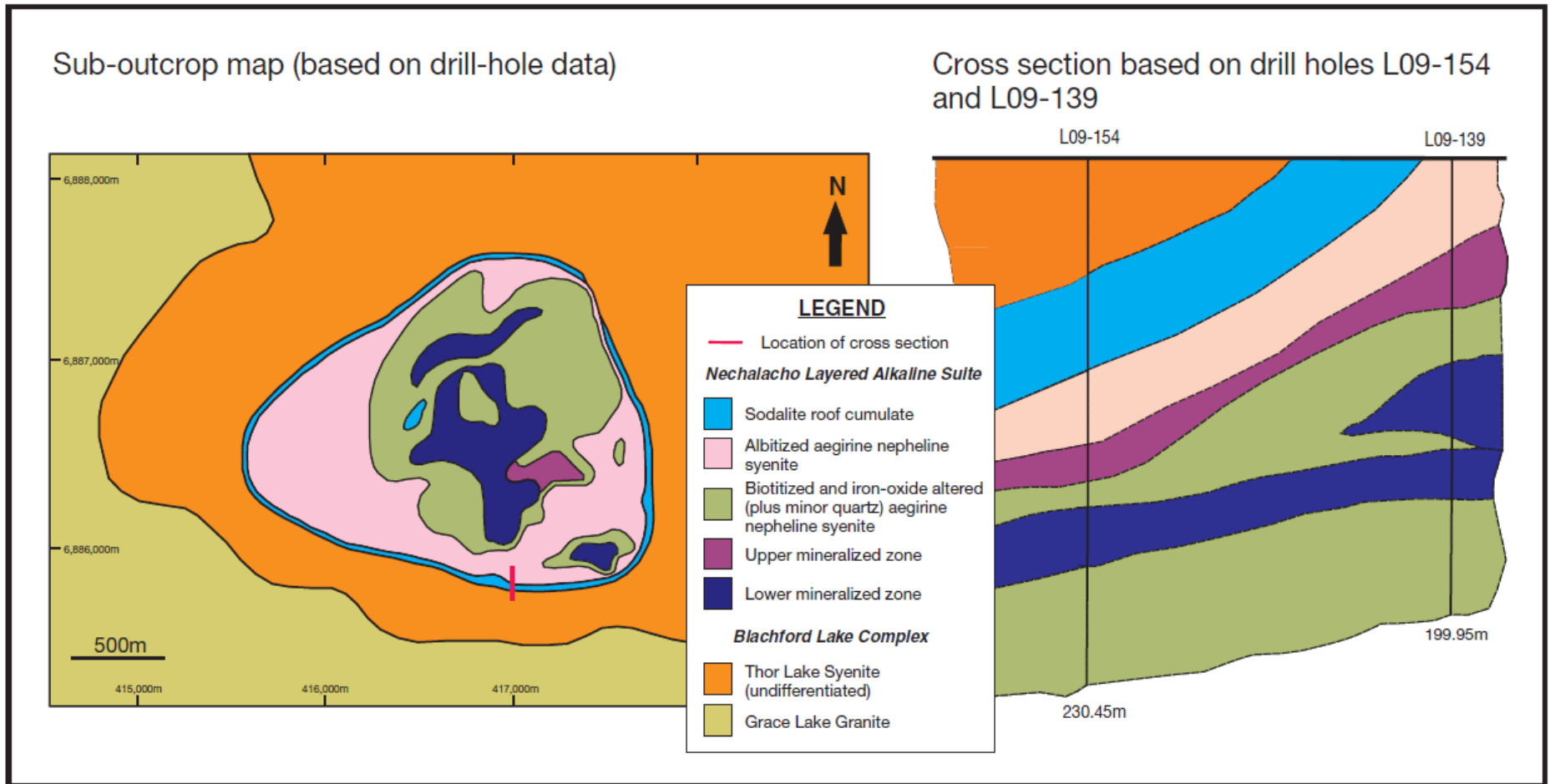


Bokan Mountain, Alaska

Bokan Mountain, Alaska



Thor Lake, NWT, Canada

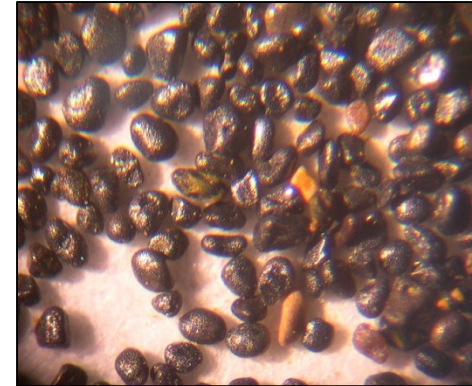


From: Sheard, E.R., Williams-Jones, A.E., Heiligmann, M., Pederson, C., and Trueman, D.L., 2012, Economic Geology, v. 107, p. 81-104.

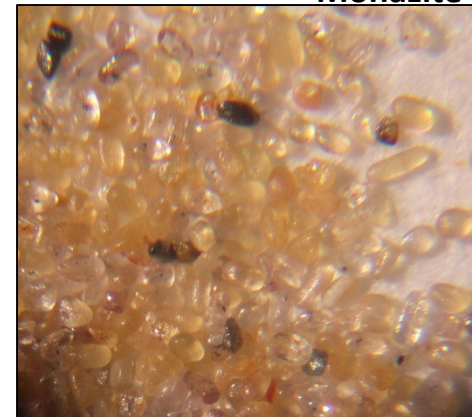
Heavy Minerals Sands (Placers)

- Heavy mineral sands are currently produced in Virginia and Florida.
- Grades average ~2% heavy minerals
 - Ilmenite 20 – 70 %
 - Zircon trace – 20%
 - Rutile, leucoxene trace – 30 %
 - Garnet, staurolite, kyanite trace – 50 %
 - Monazite trace – 15 %
- Monazite ((Ln,Th)PO₄) concentrates are currently not produced because of concerns about thorium.
- Monazite production risks: uranium & thorium

Ilmenite



Monazite

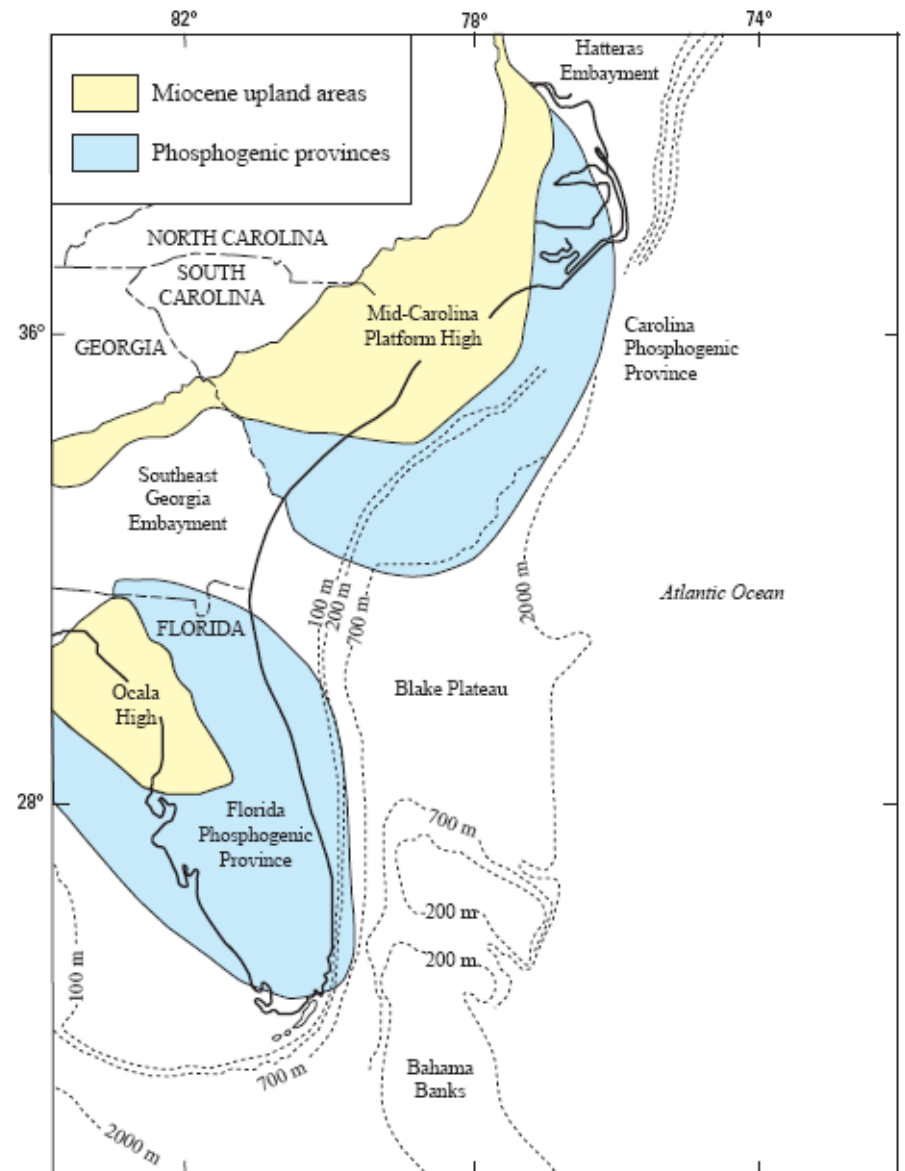


Iluka Concord Mine, Virginia



Phosphate Deposits

- Phosphate is mined in Florida (~65 %), North Carolina, Idaho and Utah
- Current potential and past production in Montana and Wyoming
- REE concentrations approach 0.5 %.
- REEs currently are not recovered.
- REEs are part of the waste stream for phosphoric acid production.
- Risks: uranium (radon), selenium



Environmental Characteristics of REE Deposits

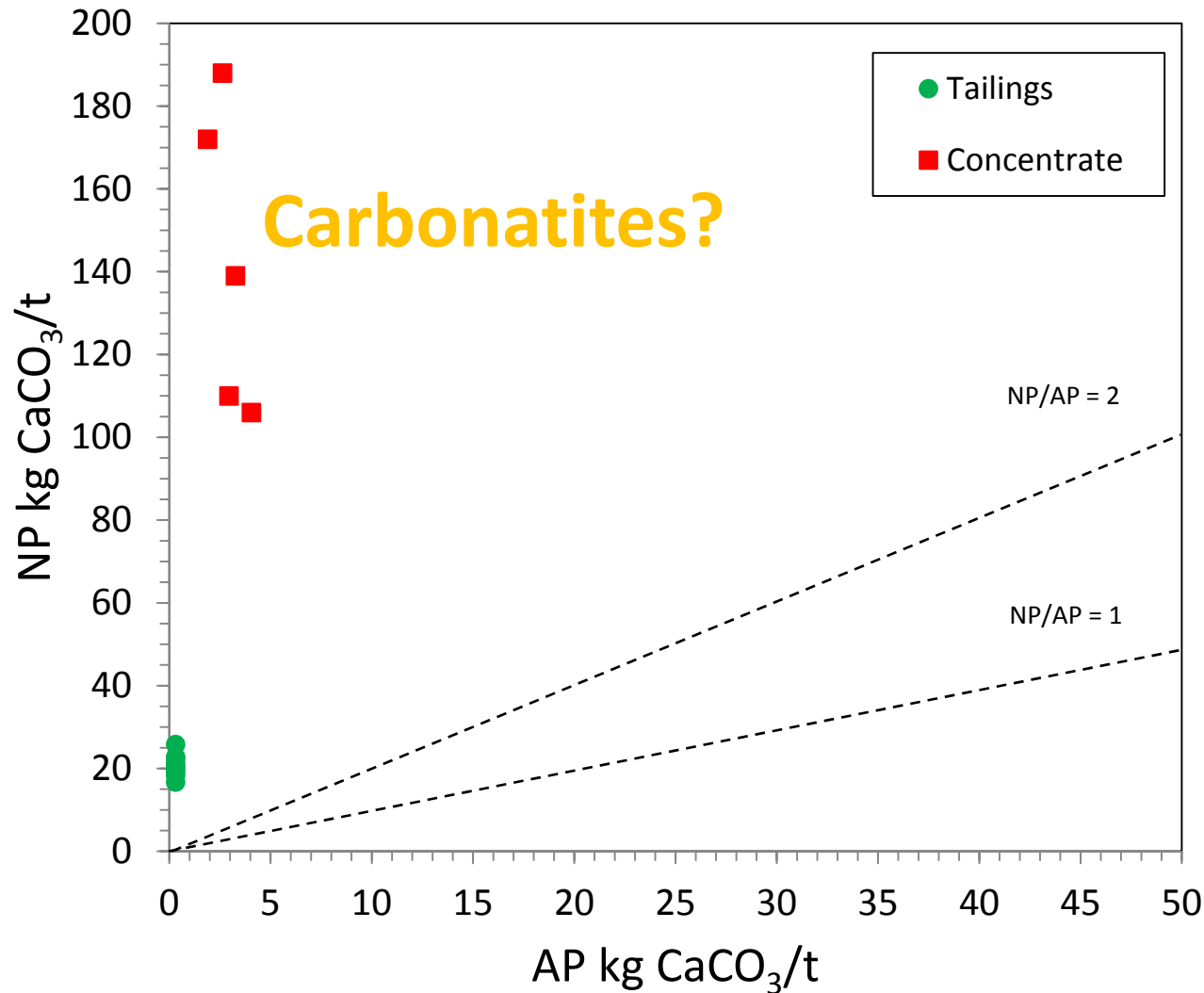
- Limited Data (Case Studies)
- Few Active, Inactive, or Abandoned Mines
 - Bayan Obo
 - Mountain Pass
 - Thor Lake (advanced exploration)
- Phosphate Fertilizer Industry
 - The Netherlands (water, sediment & soil guidelines)
 - China
- Microbial bioassay studies confirm the Free-Ion Model for lutetium (Weltie et al., 2004)
 - REE toxicity to aquatic organisms may be a more complex function of water chemistry (e.g., Wood, 1990)
- Geologic Characteristics

Rare Earth Element Minerals & Chemistry

Group-Mineral	Formula	REO Wt%	ThO ₂ Wt%	UO ₂ Wt%
Oxides				
Aeschynite	(Ln,Ca,Fe)(Ti,Nb) ₂ (O,OH) ₆			
Euxenite	(Y,Ln,Ca)(Nb,Ta,Ti) ₂ (O,OH) ₆			
Fergusonite	YNbO ₄			
Carbonates				
Bastnäsite	(Ln,Y)CO ₃ F	70 - 74	0 - 0.3	0.09
Parisite	Ca(Ln) ₂ (CO ₃) ₃ F ₂	59	0 - 0.5	0 - 0.3
Synchisite	Ca(Ln,Y)(CO ₃) ₂ F	49 - 52	1.6	
Tengerite	Y ₂ (CO ₃) ₃ •n(H ₂ O)			
Phosphates				
Apatite	(Ca,Ln) ₅ (PO ₄) ₃ (OH,F,Cl)	0 - 20		
Monazite	(Ln,Th)PO ₄	35 - 71	0 - 20	0 - 16
Xenotime	YPO ₄	52 - 67	0 - 3	0 - 5
Silicates				
Allanite	(Ln,Y,Ca) ₂ (Al,Fe ³⁺) ₂ (SiO ₄) ₃ (OH)	3 - 51	0 - 3	
Eudialyte	Na ₄ (Ca,Ce) ₂ (Fe ²⁺ ,Mn ²⁺ ,Y)ZrSi ₈ O ₂₂ (OH,Cl) ₂	1 - 10		
Thalenite	Y ₂ Si ₂ O ₇			
Zircon	(Zr,Ln)SiO ₄	0 - 0.7	0.1 - 0.8	

Ln: Lanthanide (a.k.a. REE)

Acid-Generating Potential - Thor Lake, NWT, Canada



- Despite lack of data, carbonatite-hosted REE deposits would be expected to be strongly net alkaline.

Thor Lake data: http://www.reviewboard.ca/registry/project.php?project_id=87

Rare Earth Elements: Surface Water Guidelines

	MPC* µg/L	Yellow River, Baotou, China Bayan Obo Industrial Complex			Valzinco, VA		Thor Lake, NWT, Canada
		Main Channel	Site F	Site G	VLZN-10	VLZN-3	Tailings HCT
pH		7.6 - 8.2		3.7 - 6.9	1.1	3.3	7.5
Y µg/L	6.4				370	2.0	0.183
La µg/L	10.1	0.103	140	988	340	4.6	
Ce µg/L	22.1	0.22	152	1149	870	9.5	
Pr µg/L	9.1	0.0304	16.08	294	150	1.2	
Nd µg/L	1.8	0.095	52.01	1193	650	4.7	
Sm µg/L	8.2	0.023	6.91	62.35	180	0.9	
Gd µg/L	7.1	0.028	7.22	67.33	140	0.7	
Dy µg/L	9.3	0.082	3.8	9.08	93	0.4	

*From: Maximum Permissible Concentration; Sneller et al. (2000)

Thor Lake data: http://www.reviewboard.ca/registry/project.php?project_id=87

Thor Lake Tailings & Concentrate: Leachate (HCT)

	Chronic Surface Water	Acute Surface Water	Drinking Water	Tailings
SO ₄ mg/L			250	2.9
Y µg/L	0.28	6.4		0.18
Al µg/L	750	87	200	20
As µg/L	340	150	10	0.5
Cd µg/L	0.3	2	5	0.2
Co µg/L	3.1	195		0.1
Cr µg/L	570	74	100	1.7
Cu µg/L	13	11	1,300	1.4
Fe µg/L		1,000	300	5.8
Mo µg/L	239	10,100		12.1
Ni µg/L	470	52		1.2
Pb µg/L	65	2.5	15	0.15
Sb µg/L	104	985		52
Se µg/L		5	50	<1
Sn µg/L	73.7	2680		2.2
U µg/L	1.87	33.5	30	3.9
V µg/L	19.1	284		0.04
Zn µg/L	120	120	5,000	3
Zr µg/L	54.9	982		0.71

Thor Lake data: http://www.reviewboard.ca/registry/project.php?project_id=87

Thor Lake Tailings & Concentrate: Sediment & Soil

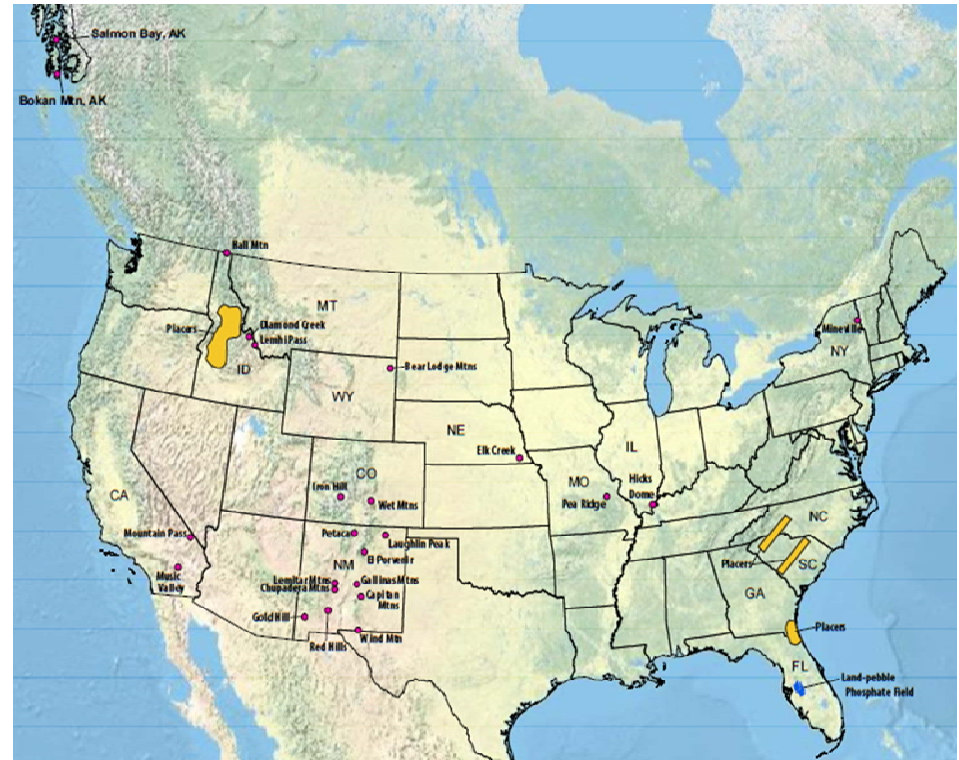
	Unit	Sediment		Soil		Thor Lake Tailings
		TEC/NC*	PEC/MPC*	Residential	Industrial	
Al	mg/kg			77,000	990,000	62,000
As	mg/kg	9.79	33	23	160	4.4
Cd	mg/kg	0.99	4.98	70	810	6
Cr	mg/kg	43.4	111	280	1,400	640
Cu	mg/kg	31.6	149	3,100	41,000	18
Mo	mg/kg			390	5,100	73
Ni	mg/kg	22.7	48.6	1,600	20,000	330
Pb	mg/kg	35.8	128	400	800	9.9
Se	mg/kg			390	5,100	3.2
U	mg/kg			230	3,100	9.9
Zn	mg/kg	121	459	23,000	310,000	68
Ce	mg/kg	256.2	18,800			1,900
Dy	mg/kg	26.2	2,200			170
Gd	mg/kg	23.2	1,800			210
La	mg/kg	83.6	4,700			850
Nd	mg/kg	43.2	7,500			970
Pr	mg/kg	66.1	5,800			250
Sm	mg/kg	30.9	2,500			210

*Negligible Concentration, Maximum Permissible Concentration; Sneller et al. (2000)

Thor Lake data: http://www.reviewboard.ca/registry/project.php?project_id=87

Geologic Summary

- Primary and byproduct deposits
 - Carbonatites
 - (Per)alkaline intrusive complexes
- Potential byproduct sources (?)
 - Phosphate deposits
 - Heavy minerals (placer) deposits
- Economic Risks
 - Estimated 1 in 10,000 showings become mines
 - Estimated 1 in 2,000 or 3,000 prospects become mines
 - Ore processing is very deposit specific because of variety of REE minerals
- In other words, the United States has potential, but do not expect a flood of new REE mines in the near future.



Environmental Summary

- Knowledge is extremely limited
 - Few mines (active, inactive, abandoned)
- Environmental risks vary among deposit types, but some risks are common to most types
- Acid-Generating Potential is Low
 - Scarce sulfide minerals
 - Carbonatites have abundant carbonate minerals

Environmental Summary (continued)

- Uranium
 - Radium: windblown dust; waterborne potential low because of low acid-generating potential
 - Radon
 - Potential for recovery
- Thorium
 - Current liability; domestic market limited
 - Future development and construction of thorium-based nuclear reactors in China, India, or France may represent future market.
- Other trace elements
 - REEs? Cd? Ni?



More work is needed!

Questions?

References

- Haxel, G.B., 2005, Ultrapotassic mafic dikes and rare earth- and barium-rich carbonatite at Mountain Pass, Mojave Desert, Southern California: Summary and field trip localities: U.S. Geological Survey Open-File Report 2005–1219, 4 p. Available at <http://pubs.usgs.gov/of/2005/1219/>.
- Haxel, G.B., Hedrick, J.B., and Orris, G.J., 2002, Rare Earth Elements—Critical Resources for High Technology: U.S. Geological Survey Fact Sheet 087-02, 96 p. Available at <http://pubs.usgs.gov/fs/2002/fs087-02/>.
- Jiang He, Chang-Wei Lu, Hong-Xi Xue, Ying Liang, Saruli Bai, Ying Sun, Li-Li Shen, Na Mi, and Qing-Yun Fan, 2010, Species and distribution of rare earth elements in the Baotou section of the Yellow River in China: *Environmental Geochemistry and Health*, v. 32, p. 45-58.
- Long, K.R., Van Gosen, B.S., Foley, N.K., and Cordier, D., 2010, The principal rare earth elements deposits of the United States—A summary of domestic deposits and a global perspective: U.S. Geological Survey Scientific Investigations Report 2010–5220, 96 p. Available at <http://pubs.usgs.gov/sir/2010/5220/>.
- Sheard, E.R., Williams-Jones, A.E., Heiligmann, M., Pederson, C., and Trueman, D.L., 2012, Controls on the concentration of zirconium, niobium, and the rare earth elements in the Thor Lake rare metal deposit, Northwest Territories, Canada: *Economic Geology*, v. 107, p. 81-104.
- Sneller, F.E.C., Kalf, D.F., Weltje, L., and van Wezel, A.P., 2000, Maximum permissible concentrations and negligible concentrations for rare earth elements (REEs): Rijksinstituut Voor Volksgezondheid En Milieu (National Institute of Public Health and the Environment), RIVM Report 601501-011, 66 p. Available at: <http://rivm.openrepository.com/rivm/bitstream/10029/9551/1/601501011.pdf>
- Verplanck, P.L., and Van Gosen, B.S., 2011, Carbonatite and alkaline intrusion-related rare earth element deposits—A deposit model: U.S. Geological Survey Open-File Report 2011–1256, 6 p. Available at <http://pubs.usgs.gov/of/2011/1256/>.
- Weltje, L., Verhoof, L.R.C.W., Verweij, W., and Hamers, T., 2004, Lutetium speciation and toxicity in a microbial bioassay: Testing the free-ion model for lanthanides: *Environmental Science and Technology*, v. 38, p. 6597-6604.
- Wood, S.A., 1990, The aqueous geochemistry of the rare-earth elements and yttrium: 1. Review of available low-temperature data for inorganic complexes and the inorganic REE speciation of natural waters: *Chemical Geology*, v. 82, p. 159-186.