

# **DETERMINATION OF UPTAKE RATES FOR VOCs IN AMBIENT AIR BY USING AXIAL TYPE THERMAL DESORPTION PASSIVE TUBES**

ORTA DOĞU TEKNİK ÜNİVERSİTESİ



MIDDLE EAST TECHNICAL UNIVERSITY

**Mihriban Yılmaz Civan, Öznur Kuntasal and Gürdal Tuncel**

ODTU  
OMETU

1

# **OUTLINE**

## **■ Introduction**

- Importance and scope of the study
- Passive Tube Theory

## **■ Material & Methods**

- Sampling Site and Duration
- Sample Preparation&Handling
- Analytical Techniques
- Performance Evaluation

## **■ Results & Discussions**

- Uptake Rate
- Meteorological Parameters

## **■ Future Works**

2

# **INTRODUCTION**

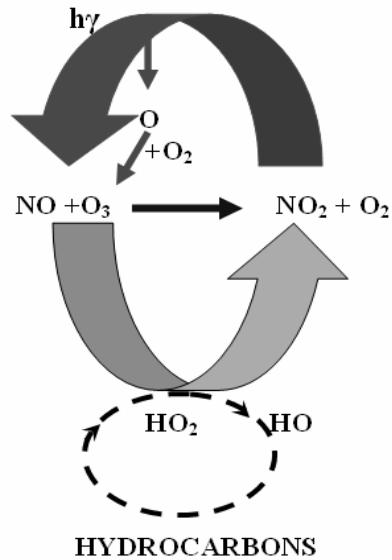
- Uptake rate for VOCs
  
- VOCs: Organic compounds having vapor pressure  $>10^{-4}$  atm @ 25 C° and 1 atm

## Introduction...

❖ VOCs are important because:

- Precursors of photochemical reactions with  $\text{NO}_x$

$\text{O}_3 \longrightarrow$  adverse health effects (iritation to noise and throat, asthma and bronchitis) and damage vegetation



## Introduction...

- toxic substances
- known or suspected carcinogen
- VOC Conc.

exhaust emmisions  
industrial process  
evaporation



## **Introduction...**

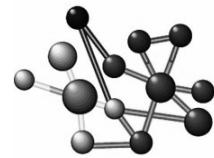
- ❖ **Sampling and Analyzing Techniques:**
  - Use near real time instrument (infrared spectrometers or portable GC)
  - sample from air using
    - ◆ Canister
    - ◆ Adsorbtion on selective sorbents (passive and active sampling)

## **Introduction...**

**PASSIVE (DIFFUSIVE) SAMPLING:**  
**taking samples by a physical process (diffusion  
and permeation)**



## Introduction...



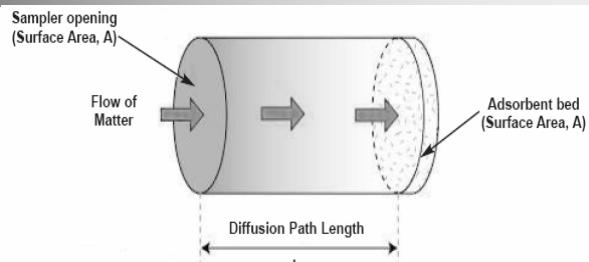
### ❖ Passive Sampling Theory:

- Adsorption of the sample onto adsorbent surface through the air movement

### ■ Fick's First Law

$$\text{Analyte Conc(ppm)} = \frac{M_a(\text{ng})}{U.R. (\text{ng / ppm} \times \text{min}) \times t(\text{min})}$$

# Introduction...



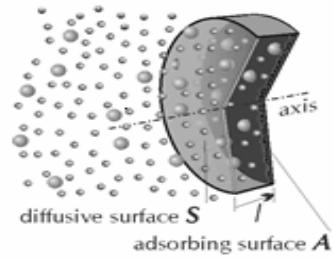
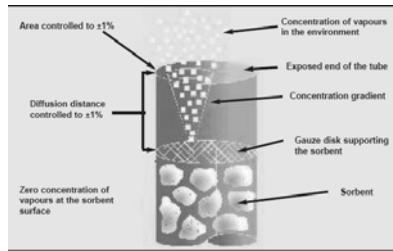
- ❖ **Uptake rate (UR):**

- Experimental U.R (exposure chamber or online GC system)**
- Ideal U.R (Diffusion Coefficient (D) is obtained from literature)**

$$Uptake\ Rate\ (ng / ppm \times min) = \frac{D(cm^2 / s) \times A(cm^2)}{L(cm)}$$

9

# Introduction...



10

# **MATERIALS & METHODS**

## **■ Sampling Site**

- Bursa Station**
  - Urban**

## **■ Sampling**

- Passive Sampling**
- Active Sampling (Hourly measurement)**

11

Continous and un interrupted data obtained

## **Sample Preparation and Handling**

Thermally desorber stainless steel passive tubes

### **Advantage:**

- ✓ not require **solvent extraction**
- ✓ have chance to **select sorbent**
- ✓ leave long term in sampling area without breakthrough
- ✓ less affected from **relative humidity**

12

## ***Sample Preparation and Handling...***

- Sorbent: **Chromosorb 106 by Supelco**
- **Characteristics:**
  - suitable analyte volatility range for VOC in concern
  - **Hydrophobic**
  - analyte volatility range boiling point **50 °C-200 °C**
  - maximum temperature: **250 °C**
  - specific surface area : **750 m<sup>2</sup>/g**

13

# Sampling Methodology

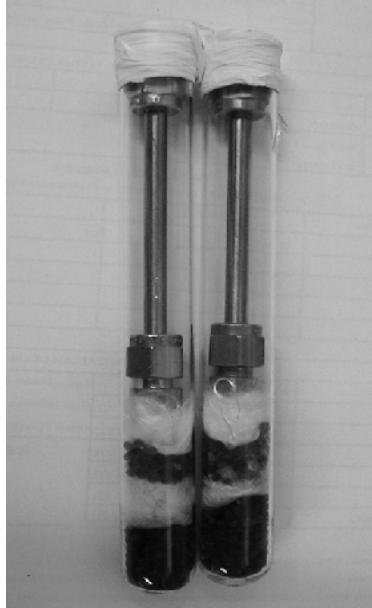


- **Condition** of tubes
- **Swagelock**-type screw caps with combined PTFE ferrule
- Sealed glass tubes filled with **silica gel and charcoal** at the bottom
- **Deep-freeze** in the laboratory
- **Glass sealed** jar filled with activated charcoal

## **Sampling Methodology ...**

- European Standard (EN) 13528
- Shelters** made from aluminum
- Replacing the sampling cap with **diffusion part**
- placed between **1.6 - 2.0 m** from ground level
- Field /Lab Blanks**

15



## **Methodology...**

- **Analysis of collected samples on sorbents**
- **Organics (VOCs) → GC-FID-Unity TD**

## **Methodology...**

- ❖ **GC-FID coupled with Unity Air Server**
  - **Markes Unity-Air Server Thermal Desorber**
  - **HP 6890 Dual Column GC coupled to FID with Dean Switch System**

## Methodology... (GC-FID)



19

## Methodology... (GC-FID)



20

## **Methodology...**

**□ Calibration **gas** standard:**

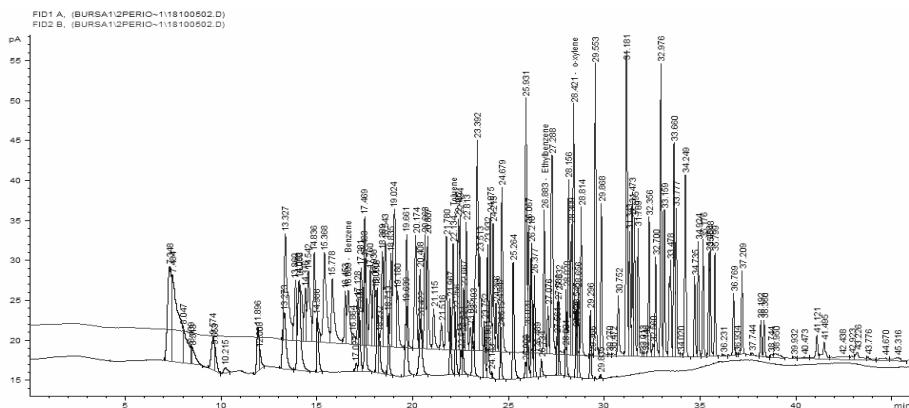
148 VOCs (**C<sub>2</sub>-C<sub>12</sub>**) from AAQD of  
Environment Canada

✓ **5 point calibration**

**□ Sorrogate Standard:**

1-Bromoflurobenzene

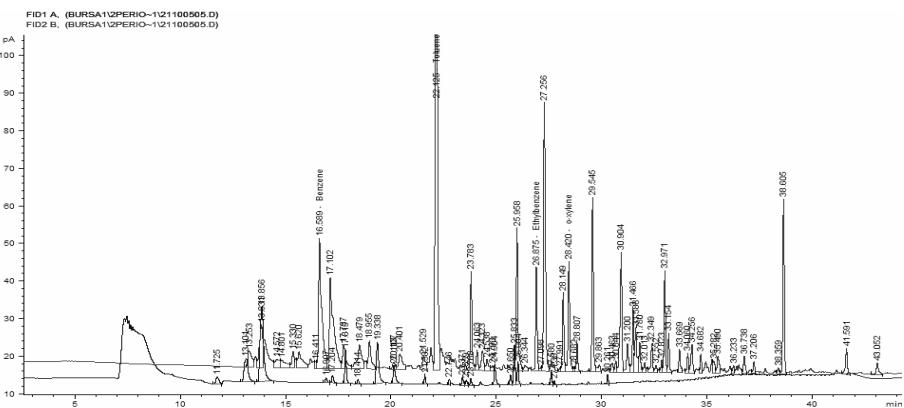
# Methodology...



Typical Calibration Standard Chromatogram

22

# Methodology...



Typical Sample Chromatogram

23

## ***Performance Evaluation***

- ❖ **Detection Limit**

$0.21 \text{ } \mu\text{g m}^{-3}$  ( $0.08 \text{ } \mu\text{g m}^{-3}$  -  $0.31 \text{ } \mu\text{g m}^{-3}$ )

- ❖ **Recovery/Desorption Efficiency**

93.6% (80 - 100%)

- Precision**

7.7% (2.6 - 15.7%)

- ❖ **Sampling Stability ??**



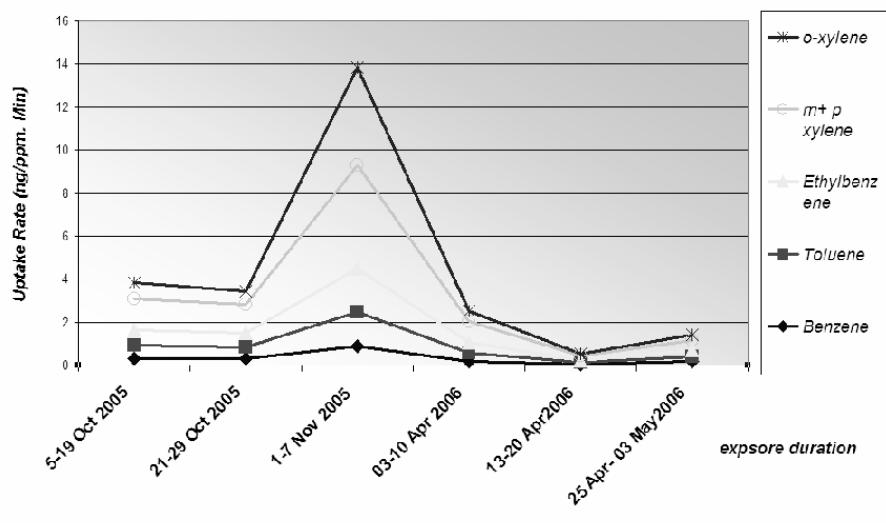
25

# **RESULT**

<b>Exposure Date</b>	<b>Number of active samples</b>	<b>Number of passive tubes</b>
<b>5-19 Oct.2005</b>	258 (77%)	3
<b>21-29 Oct.2005</b>	173 (90%)	5
<b>1-7 Nov.2005</b>	84 (58%)	4
<b>03-10 Apr.2006</b>	144 (86%)	6
<b>13-20 Apr2006</b>	152 (90%)	4
<b>25.Apr-03 May.2006</b>	181 (100%)	3

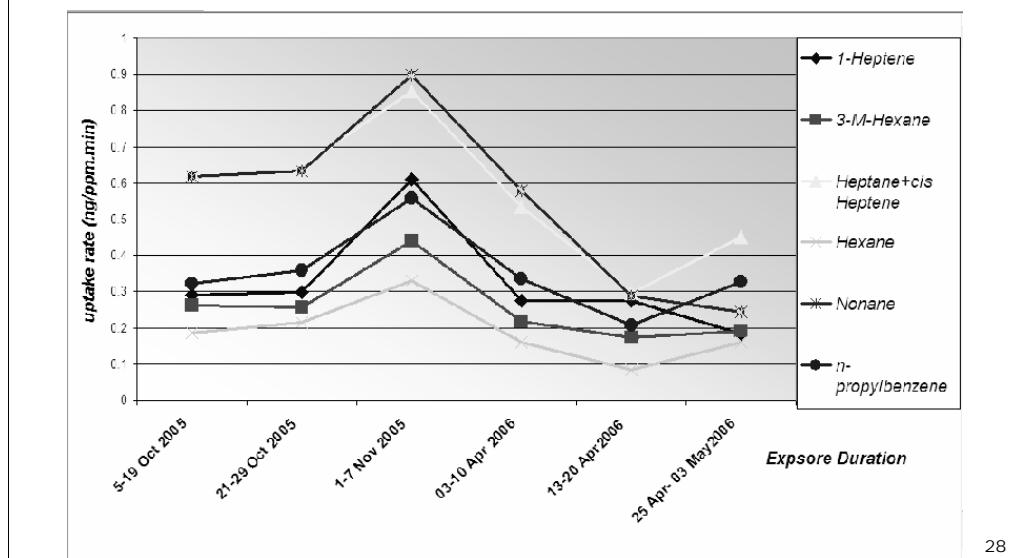
26

## Uptake Rates for BTEX



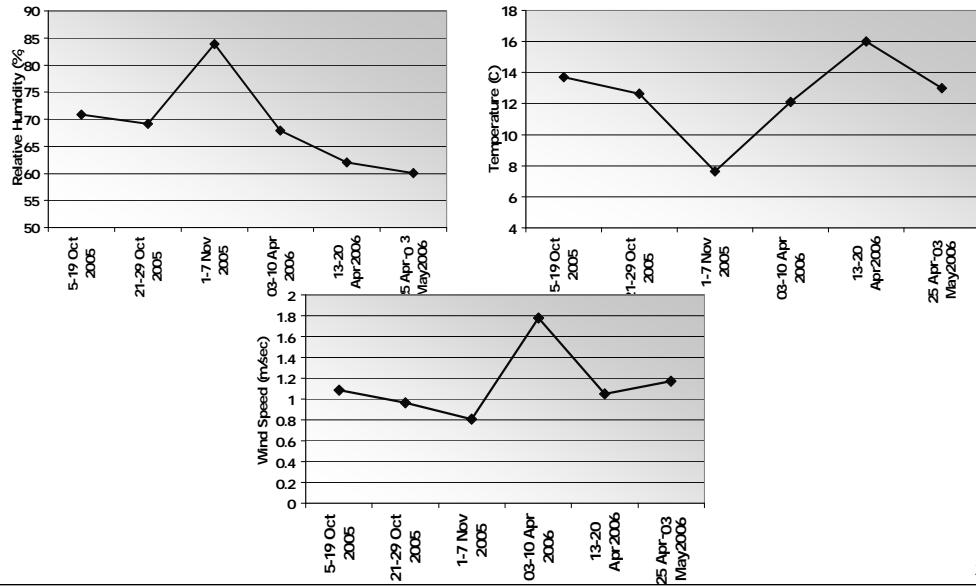
27

## Uptake Rates for NMHCs



28

## Meteorological Parameters:



29

## UPTAKE RATE

COMPOUND	UPTAKE RATE (ng ppb <sup>-1</sup> min <sup>-1</sup> )
Benzene	0.323
Toluene	0.562
Ethylbenzene	0.718
m&p-xylene	1.535
o-xylene	1.123

30

## UPTAKE RATE...

COMPOUND	UPTAKE RATE (ng ppb <sup>-1</sup> min <sup>-1</sup> )
1-Heptene	0.3219
3-M-Hexane	0.257
Heptane& cis-3-Heptene	0.568
n-Hexane	0.189
n-Nonane	0.543
n-Probylbenzene	0.351

31

## **Future Works...**

- Long term sampling with canister
- Meteorological parameter
- QA/QC complete

32